



Structure and flexibility of the thermophilic cold-shock protein of *Thermus aquaticus*

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ABSTRACT

The thermophilic bacterium *Thermus aquaticus* is a well-known source of Taq polymerase. Here, we studied the structure and dynamics of the *T. aquaticus* cold-shock protein (*Ta*-Csp) to better understand its thermostability using NMR spectroscopy. We found that *Ta*-Csp has a five-stranded β -barrel structure with five salt bridges which are important for more rigid structure and a higher melting temperature (76 °C) of *Ta*-Csp compared to mesophilic and psychrophilic Csps. Microsecond to millisecond time scale exchange processes occur only at the β 1– β 2 surface region of the nucleic acid binding site with an average conformational exchange rate constant of 674 s^{−1}. The results imply that thermophilic *Ta*-Csp has a more rigid structure and may not need high structural flexibility to accommodate nucleic acids upon cold shock compared to its mesophile and psychrophile counterparts.

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1. Introduction

Many organisms have the ability to respond to changes in the environment, such as changes in temperature, pressure, osmotic conditions, and oxygen availability. Bacteria have evolved specific physiological mechanisms to help them adapt to temperatures lower than their optimal growth temperatures. The adaptation of bacteria to minimal growth temperature is reflected by changes in the growth rate, levels of fatty acid saturation, and the regulation of DNA, RNA, and protein synthesis in the cells [1]. At minimal growth temperature, synthesis of most proteins is downregulated, while that of cold-shock proteins (Csps) is upregulated [2].

Csps are found in a wide range of bacteria, including psychrophilic, mesophilic, thermophilic, and hyperthermophilic species. They share a highly conserved sequence and structure. Csps function as RNA chaperones, assisting the unfolding of mRNAs [3,4]. The three-dimensional structures of several Csps have been resolved by X-ray crystallography and nuclear magnetic resonance (NMR) spectroscopy. These include the following: mesophilic Csps, such as those of *Escherichia coli* (*Ec*-CspA) [5,6], *Bacillus subtilis* (*Bs*-CspB) [7–9], *Salmonella typhimurium* (*St*-Csp) [10], *Mycobacterium tuberculosis* (*MTB*-Csp) [11], and *Rickettsia rickettsia* (*Rr*-Csp) [12]; thermophilic Csps, such as the *Bacillus caldolyticus* Csp (*Bc*-Csp) [13,14] and *Thermotoga maritima* Csp (*Tm*-Csp) [15]; and a psychrophilic Csp, the *Listeria monocytogenes* CspA (*Lm*-CspA) [16].

All of these Csps contain five antiparallel β -strands with an oligonucleotide-binding fold. Aromatic residues located in two of these conserved nucleic acid binding motifs, ribonucleoprotein (RNP) 1 and RNP 2, facilitate binding with single-stranded nucleic acids.

Csps from thermophilic bacteria with known 3D structures are *Bc*-Csp from the *B. caldolyticus* and *Tm*-Csp from the hyperthermophilic bacteria *T. maritima*. Both Csps have higher thermal stability than their mesophile and psychrophile counterparts. However, motional properties of thermophilic Csps have not been studied by spin relaxation experiment, yet.

Thermus aquaticus is a species of thermophilic bacteria that thrives at 70 °C, but can survive at temperatures ranging from 50 °C to 80 °C. It is the source of the Taq DNA polymerase widely used for DNA amplification in the polymerase chain reaction technique. Here, we determined the solution structures of free *Ta*-Csp and the single-stranded nucleic acid [heptathymidine (dT₇)]-bound complex *Ta*-Csp-dT₇. For the first time, we investigated the structural flexibilities of *Ta*-Csp to understand the origin of the high thermostability of thermophilic Csps using spin relaxation experiment.

2. Materials and methods

2.1. Protein expression, isotopic enrichment, and purification

Expression of *Ta*-Csp, isotopic enrichment, and purification was performed as described previously [16]. Concentrated protein was purified using size-exclusion and hydrophobic-interaction

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chromatography. The protein yield from 1 L of cell culture was generally between 0.5 and 1 mg.

2.2. Circular dichroism measurement of Ta-Csp

Circular dichroism (CD) measurements were performed in a J810 spectropolarimeter (Jasco) using a cell with a 1-mm path length. The CD spectra of 50 μ M Ta-Csp were measured at 0.1 nm intervals from 190 nm to 250 nm in 50 mM potassium phosphate buffer (pH 6.0) containing 100 mM KCl and 0.1 mM EDTA at 25 °C; data from 10 scans were averaged. CD data were expressed as the mean residue ellipticity (θ) in degrees square centimeters per decimole. Protein T_m values were determined from a series of CD spectra obtained at temperatures ranging from 15 °C to 100 °C.

2.3. NMR experiments and structure determination

All NMR experiments were performed as described previously [16] at 25 °C on a Bruker Avance 500 and 800 MHz spectrometer at the Korea Basic Science Institute. Nuclear Overhauser effect (NOE) distance constraints were measured from 3D ^1H – ^{15}N and ^1H – ^{13}C nuclear Overhauser effect spectroscopy–heteronuclear single-quantum coherence (NOESY–HSQC) spectra acquired with an 800 MHz Bruker Avance NMR spectrometer equipped with a ^1H –(^{13}C , ^{15}N) triple-resonance probe using a mixing time of 120 ms [17–19]. NMR spectra were processed with NMRPipe [20] and decoded with Sparky [21]. Structure calculations were performed using the CYANA 2.1 [22] together with dihedral (ϕ and ψ) constraints using the TALOS [23]. Final coordinates and NOE constraints have been deposited in the Protein Data Bank (PDB) as entries 2MO1 (Ta-Csp-dT₇) and 2MO0 (free Ta-Csp).

2.4. NMR investigation of backbone dynamics

We performed NMR spin-relaxation experiments on free Ta-Csp and dT₇-bound Ta-Csp (at a molar ratio of Ta-Csp to dT₇ of 1:1) and analyzed R_1 , R_2 , and heteronuclear NOE (hNOE) NMR spectra acquired on a Bruker Avance 500 MHz spectrometer [24]. Longitudinal (R_1) spin-relaxation rates were obtained, and relaxation delays were 0.00200 (measured twice), 0.0450, 0.100, 0.200, 0.315 (measured twice), 0.550, 0.800, and 1.00 s. The transverse (R_2) relaxation rate was measured with relaxation delays of 0 (measured twice), 0.0166, 0.0333, 0.0499 (measured twice), 0.0666, 0.116, 0.183, and 0.283 s. The heteronuclear cross-relaxation rate was obtained from NOE experiments by interleaving pulse sequences with and without proton saturation. The recycle delay and proton saturation time in hNOE measurements were 4.5 and 3.0 s, respectively. All relaxation spectra were acquired as described previously [16]. The data were processed with NMRPipe [20] and visualized using Sparky [21]. R_1 , R_2 , and hNOEs rates were determined as described in our previous paper [16,25,26]. Constant-time relaxation-compensated Carr–Purcell–Meiboom–Gill (CPMG) experiments [27] were performed at 500 MHz and 800 MHz spectrometer as described previously [16]. The intensities of cross-peaks were then converted into decay rates, R_2^{eff} , for a given field strength, ν_{cp} [28]. CPMG dispersion profiles were fit to a two-state exchange model using CATIA program from F. Hansen (<http://pound.med.utoronto.ca/~flemming/catia>).

2.5. Fluorescence quenching

Quenching of the intrinsic fluorescence of W8 in Ta-Csp upon binding to nucleic acids was used to determine binding constants. The binding constant of oligonucleotides was measured on a RF-5301PC spectrofluorophotometer. Ta-Csp protein (final concentration, 10 μ M) was added to 50 mM potassium phosphate buffer

containing 100 mM KCl and 0.1 mM EDTA at pH 6.0 at a final protein-to-oligonucleotide ratio of 1:10. The sample was placed in a 2 ml thermostatically controlled cuvette, and excitation and emission path lengths were set at 10 nm. The sample was excited at 290 nm, and emission spectra were recorded for light-scattering effects from 300 nm to 500 nm. Dissociation constants (K_d) were estimated as described previously [16].

3. Results

3.1. Thermostability of Ta-Csp and Ta-Csp-dT₇ determined through CD experiments

Csps from different species, ranging from psychrophilic to thermophilic bacteria, share high sequence as well as structural similarity. Ta-Csp showed 59% amino acid sequence identity with Bc-Csp, 56% with Tm-Csp, and 54% with Bs-CspB. Despite these structural similarities, they display dramatic differences in thermostability [29]. In order to better understand the cold-shock response in bacteria, we explored structural characteristics that may potentially account for these differences in thermostability among Csps.

The secondary structures of Ta-Csp and Ta-Csp-dT₇ were investigated by CD measurements. As shown in Fig. 1A, CD spectra of Ta-Csp and Ta-Csp-dT₇ are very similar, showing a maximum at 200 nm and a minimum at 212 nm, characteristic of a β -sheet structure. Analysis of the thermal denaturation curves obtained at temperatures ranging from 15 °C to 100 °C showed the T_m of Ta-Csp to be 76 °C, indicating that Ta-Csp is much more thermostable than psychrophilic and mesophilic Csps.

3.2. Solution structures of Ta-Csp and Ta-Csp-dT₇

The three-dimensional structures of free Ta-Csp and Ta-Csp-dT₇ were calculated based on NMR constraints; NMR structural statistics for these calculations are summarized in Table 1. We included 979 and 978 conformational constraints in the structure calculation of free Ta-Csp and Ta-Csp-dT₇, respectively. The structures revealed that free Ta-Csp forms a Greek-key topology, a closed β -barrel consisting of 5 β -strands; 2 residues in these β -sheets (T5 and K64) form β -bulges, which break strand β_1 into β_1' and β_1'' and strand β_5 into β_5' and β_5'' . As a result, two β -sheets surfaces— β -sheets 1 and 2—are formed by strands β_1'' – β_2 – β_3 and β_1' – β_4 – β_5' – β_5'' . Five well-defined β -strands were identified in free Ta-Csp: β_1 (K2–N10), β_2 (Y15–Q20), β_3 (D25–H29), β_4 (H46–G54), and β_5 (P60–R67). They can be arranged into 2 antiparallel β -pleated sheets stabilized by hydrophobic interactions and hydrogen bond networks within the β -sheets. The side chain groups of V6, F9, I18, V26, V28, I33, L41, V47, F49, V51, A62, and V65 point to the interior of the β -barrel structure in Ta-Csp, forming a hydrophobic core. Flexible loop 3 is approximately 16 residues long, and loop 4 encompasses 5 residues from R55 to G59.

The structure of Ta-Csp-dT₇ is almost invariant compared to that of the free form, having exactly the same five β strands. As shown in Fig. 1B and C both the Ta-Csp structure and the structure of the complex displayed ionic interactions between various residues: E21–K3–E48–R66, K7–D25, and E52–K63. Fig. 1D shows the solvent-exposed hydrophobic residues such as W8, Y15, F17, F27, V28, F30, and F38 as well as the hydrophilic residues which form a cluster of protein surfaces and are important for binding to single-stranded nucleic acids [8,9]. The backbone root mean square deviations (RMSDs) between free Ta-Csp and dT₇-bound Ta-Csp for the 5 β strands and the 4 loops were measured to be 0.22 Å (β_1), 0.30 Å (β_2), 0.31 Å (β_3), 0.27 Å (β_4), 0.15 Å (β_5), 0.16 Å (loop 1), 1.10 Å (loop 2), 1.18 Å (loop 3), and 0.25 Å (loop 4). The backbone RMSD relative to the mean structure for all residues was

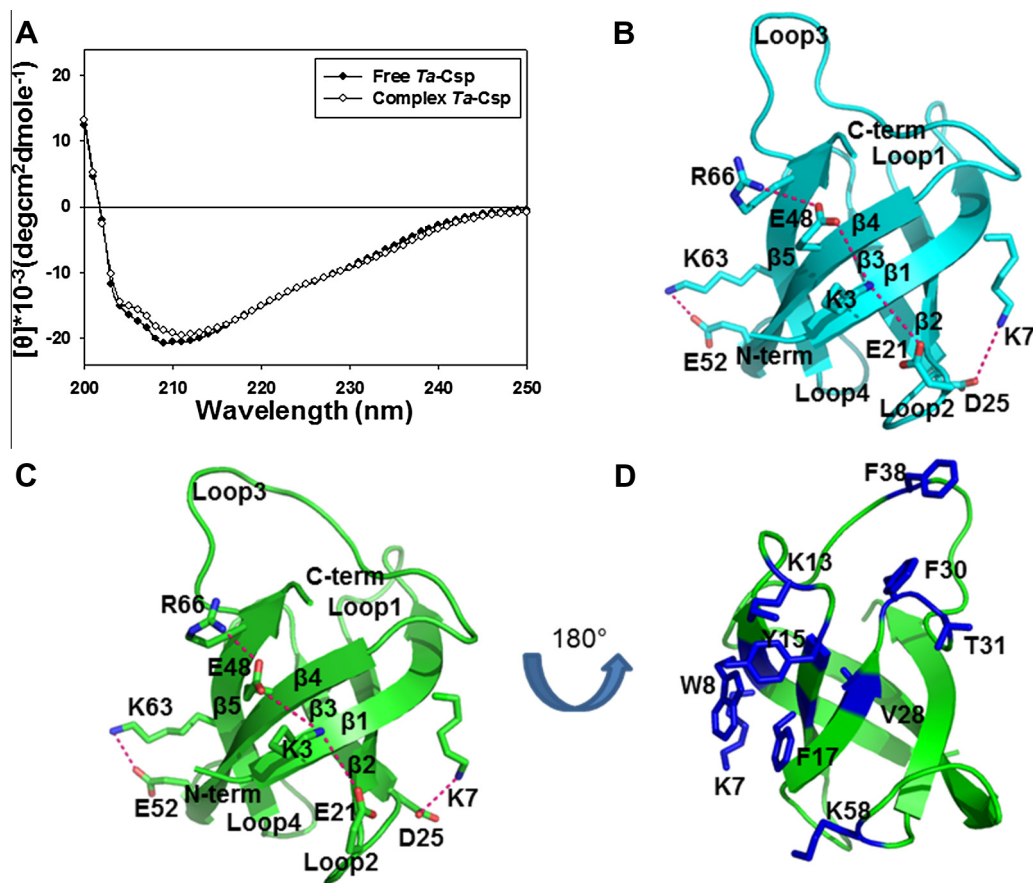


Fig. 1. (A) CD spectra of *Ta-Csp* (filled circles) and *Ta-Csp-dT7* (open circles) in 50 mM potassium phosphate containing 100 mM KCl and 0.1 mM EDTA (pH 6.0) at 25 °C. (B) Average NMR structures of free *Ta-Csp* (cyan) and (C) *Ta-Csp-dT7* (green) determined using NMR spectroscopy. Hydrogen bonds are represented by dotted lines. (D) Structure of *Ta-Csp-dT7* showing the side chains of dT7 binding residues in blue. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Table 1
Structural statistics and mean pairwise root mean squared deviations for the 20 lowest-energy structures of *Ta-Csp* and *Ta-Csp-dT7*.

Distance restraints	Free <i>Ta-Csp</i>	Complexed <i>Ta-Csp</i>
Total	979	968
Intraresidual and sequential	510	500
Medium-range	105	104
Long-range	364	364
Dihedral angle restraints	117	117
Hydrogen bonds	28	28
Rmsd of the backbone all residues (Å)	0.43 ± 0.20	0.39 ± 0.10
Rmsd from idealized geometry		
Bonds (Å)	0.0035 ± 0.00001	0.0035 ± 0.00002
Angle (°)	0.47 ± 0.019	0.46 ± 0.036
Water refinement energies (kcal mol ⁻¹)		
E_{NOE}	11.16 ± 11	14.16 ± 1
E_{cdih}	1.12 ± 0.006	1.71 ± 0.01
E_{elec}	-2653 ± 23	-2884 ± 24
E_{vdw}	-321 ± 3	-261 ± 6
E_{total}	-2536 ± 21	-2661 ± 22
Average violations		
Distance violation (0.50 Å < d)	0	0
Angle violation ($\theta > 5^\circ$)	0	0
Ramachandran plot statistics (%)		
Residues in most favored regions	75.7	77.5
Residues in additional allowed regions	23.4	22.4
Residues in generously allowed regions	0.9	0.1
Residues in disallowed regions	0	0

0.43 ± 0.20 Å for free *Ta-Csp* and 0.39 ± 0.10 Å for *Ta-Csp-dT7* (Table 1). In the β -sheet-ordered regions, backbone RMSD relative to the mean structure was 0.27 ± 0.02 Å for free *Ta-Csp* and 0.28 ± 0.01 Å for *Ta-Csp-dT7*.

3.3. Chemical shift perturbations of *Ta-Csp* upon dT7 binding

In order to investigate local structural rearrangements upon a ligand binding to the protein, chemical shift variations were analyzed by comparing the ¹⁵N-¹H HSQC spectra of free *Ta-Csp* and *Ta-Csp-dT7*. Fig. 2A shows the overlay of the two-dimensional ¹⁵N-¹H HSQC spectra of *Ta-Csp* and dT7-bound *Ta-Csp* at a *Ta-Csp*-to-dT7 molar ratio of 1:1. Blue lines in Fig. 2A and B indicate residues showing larger chemical shift variations upon dT7 binding more than 0.25 ppm. Large changes in chemical shift imply that most residues facilitate the binding of dT7 through direct interaction between the base or phosphate group of dT7 and the binding site, and also through conformational rearrangements of the flexible loop region away from the binding interface such as I33, L41 and G56. These binding residues are exactly the same to the nucleic acid binding residues in NMR and X-ray structures of *Bs-CspB* [8,9]. Chemical shift perturbation was seen at the imidazole ring NH in W8 (from 10.43 ppm to 9.73 ppm), too. The peak traces of K7 and Y15 of *Ta-Csp* which side chains form close contacts with dT7 are shown in the box of ¹H-¹⁵N HSQC spectra. The NMR signals gradually shifted in a single direction with the addition of dT7, indicating that the exchange rate of dT7 between the free and the bound states is faster than the NMR time scale.

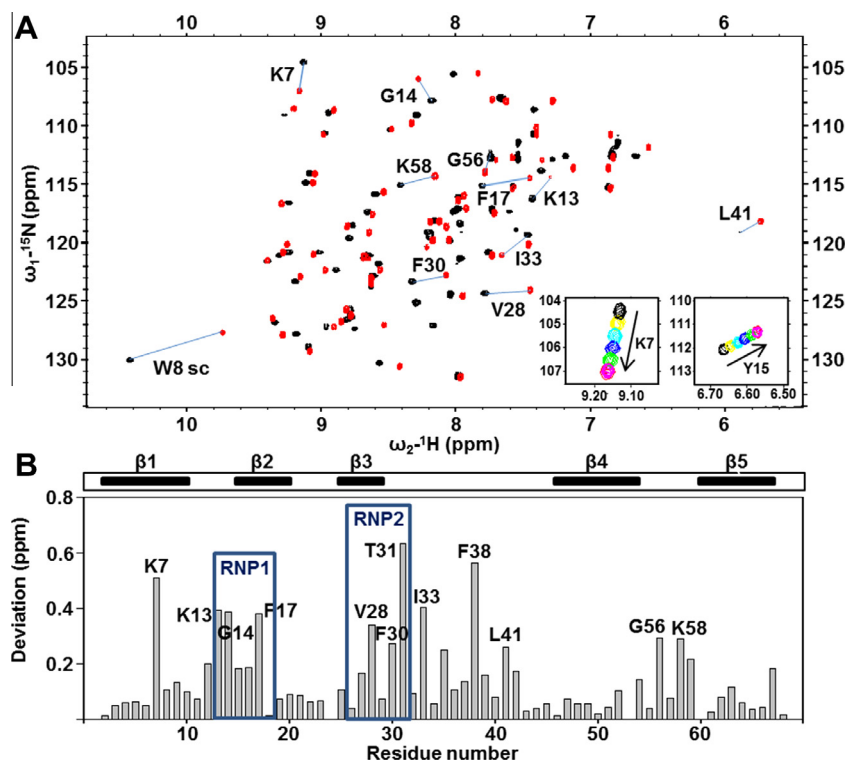


Fig. 2. (A) Overlay of ^1H – ^{15}N HSQC spectra of free *Ta*-Csp (black) and *Ta*-Csp–dT₇ (red) tested at a 1:1 molar ratio of *Ta*-Csp to dT₇. Protein samples were prepared in 50 mM potassium phosphate buffer containing 100 mM KCl and 0.1 mM EDTA at pH 6.0. Peak traces of K7 and Y15 for the dT₇ titration shown in the box are the signal changes of *Ta*-Csp after titration with dT₇ at ratios of (a) 0:1 (black), (b) 0.1:1 (yellow), (c) 0.3:1 (sky blue), (d) 0.5:1 (blue), (e) 0.7:1 (green), (f) 1:1 (red), and (g) 1.2:1 (magenta). (B) Chemical shift perturbation upon addition of dT₇. Weighted average of the ^{15}N and ^1H chemical shift perturbations ($\Delta\delta = \sqrt{(\delta\text{H}^2 + 0.2(\delta\text{N}^2))}$) of *Ta*-Csp upon complex formation with dT₇. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

3.3.1. ^{15}N NMR relaxation study

The results of spin-relaxation experiments with free *Ta*-Csp and dT₇-bound *Ta*-Csp were compared and are shown in Fig. 3. The average R_1 , R_2 , and hNOE values of free *Ta*-Csp were $2.42 \pm 0.045 \text{ s}^{-1}$, $6.78 \pm 0.53 \text{ s}^{-1}$, and 0.56 ± 0.017 units, respectively; the corresponding values for dT₇-bound *Ta*-Csp were $2.55 \pm 0.051 \text{ s}^{-1}$, $6.03 \pm 0.21 \text{ s}^{-1}$, and 0.68 ± 0.027 units, respectively. The average R_2/R_1 ratio for free *Ta*-Csp was 2.80 ± 0.38 , whereas that for bound *Ta*-Csp was 2.36 ± 0.15 , implying that both free *Ta*-Csp and the bound form exist as monomers. Large R_2 values ($>10 \text{ s}^{-1}$) for the backbone N–H of K7 and W8 could be due to the conformational exchange (R_{ex}) of the protein on microsecond to millisecond time scales. Conformational exchanges of K7 ($10.77 \rightarrow 7.77 \text{ s}^{-1}$), and W8 ($15.47 \rightarrow 6.50 \text{ s}^{-1}$) at the dT₇ binding site decreases dramatically upon binding, resulting in highly populated bound conformations.

Although there are not many structural differences between free and complex structures, the binding of dT₇ to *Ta*-Csp increased the hNOE values in overall structures. The average hNOE value was 0.56 for free *Ta*-Csp, which increased to 0.68 for the dT₇-bound form. The average hNOE values at loop regions were 0.590 (loop 1), 0.528 (loop 2), 0.513 (loop 3), and 0.487 (loop 4). The binding of dT₇ to *Ta*-Csp dramatically increased the hNOE values of these loop regions to 0.758, 0.573, 0.634, and 0.641, respectively. The average hNOE values at β -sheet regions also increased: 0.624 (β_1), 0.643 (β_2), 0.620 (β_3), 0.574 (β_4) and 0.585 (β_5). The binding of dT₇ to *Ta*-Csp increased the hNOE values of these regions to 0.719 (β_1), 0.761 (β_2), 0.716 (β_3), 0.669 (β_4), and 0.713 (β_5). Therefore, upon dT₇ binding, the fast time scale motions in the *Ta*-Csp structure decrease and the structure of *Ta*-Csp becomes rigid.

Microsecond to millisecond motions in *Ta*-Csp were investigated by a CPMG experiment on *Ta*-Csp and its bound form. Fig. 4A and B show the differences in R_2^{eff} values as a function of amino acid sequence, ΔR_2^{eff} (v_{CP}), revealing that dT₇ binding reduced slow time scale motion, dramatically decreasing the R_{ex} values of the binding epitope and stabilizing the structure of *Ta*-Csp. In particular, the ΔR_2^{eff} values of K7 ($20.19 \rightarrow 7.05 \text{ s}^{-1}$), W8 ($24.02 \rightarrow 4.36 \text{ s}^{-1}$), Y15 ($15.13 \rightarrow 4.19 \text{ s}^{-1}$) and F17 ($13.93 \rightarrow 3.39 \text{ s}^{-1}$) dramatically decreased upon dT₇ binding. The relaxation dispersion data at 500 MHz NMR and 800 MHz NMR for W8 are shown in Fig. 4C. W8 located at the binding site underwent chemical exchange on a microsecond to millisecond time scale in free *Ta*-Csp, while the relaxation dispersions of these residues disappeared in dT₇-bound *Ta*-Csp. CPMG dispersion profiles were fit to a two-state exchange model and the rates for K7 ($690 \pm 55.7 \text{ s}^{-1}$), W8 ($841 \pm 67.3 \text{ s}^{-1}$), Y15 ($616 \pm 53.1 \text{ s}^{-1}$), and F17 ($547 \pm 45.2 \text{ s}^{-1}$) results in an average exchange rate constant of $674 \pm 55.3 \text{ s}^{-1}$ for these residues at the β_1 – β_2 surface of the nucleic acid binding site.

4. Discussion

The three-dimensional structures and the primary sequences of CspA from psychrophilic, mesophilic, thermophilic, and hyperthermophilic bacteria are highly similar and yet the CspA exhibit a diverse range of melting points. Previously, we investigated the dynamic properties and structure of the psychrophilic *Lm*-CspA using NMR spectroscopy, and showed that *Lm*-CspA is less thermostable ($\sim 40^\circ\text{C}$) than mesophilic CspA, and exhibits high structural flexibility [16]. The structure of *Lm*-CspA contains two salt bridges,

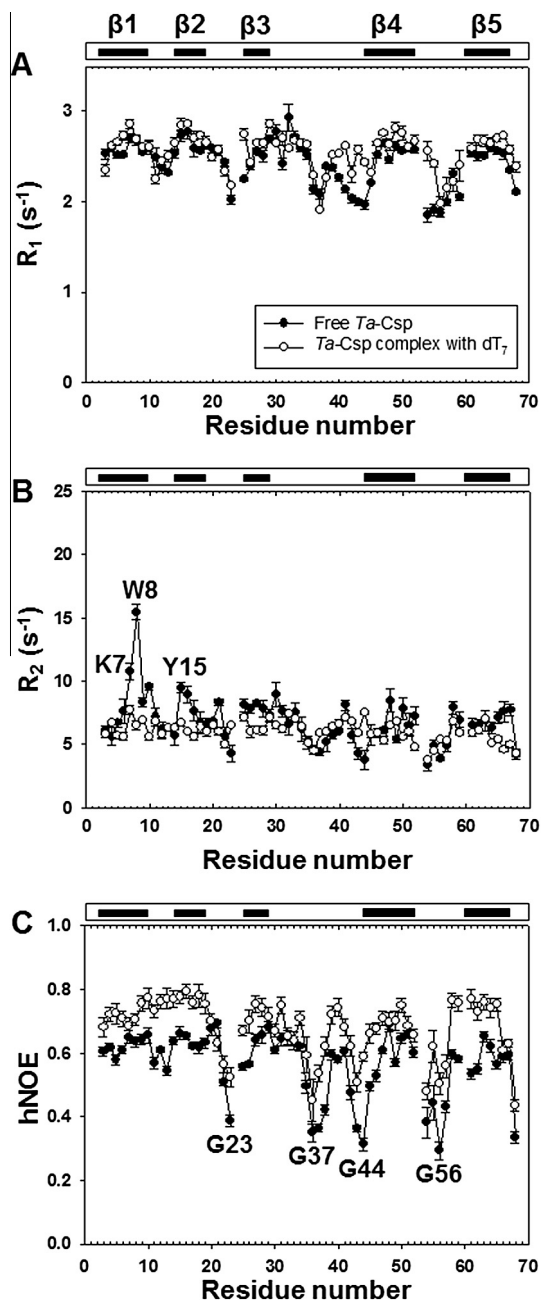


Fig. 3. Comparison of (A) R_1 , (B) R_2 , and (C) hNOE values for free *Ta*-Csp (filled circles) and *Ta*-Csp-dT₇ (open circles) at pH 6.0, at a molar ratio of *Ta*-Csp to dT₇ of 1:1. Solid bars represent the five β-strands of *Ta*-Csp.

E2–R20 and K7–D25, and 10 hydrophobic residues (V6, F9, I18, V26, V28, L41, V47, F49, V51, and V63) forming the hydrophobic core. The mesophilic *Bs*-CspB with a T_m (~52 °C) [7] also contains two salt bridges, K7–D25 and K5–E19, but has two more hydrophobic core residues than the psychrophilic *Lm*-CspA. These hydrophobic interactions in mesophilic Csps result in a higher T_m compared to that of psychrophilic Csp. The thermophilic *Ta*-Csp with a T_m (~76 °C) includes 12 hydrophobic core residues, the same number as in mesophilic Csps. However, *Ta*-Csp has more intramolecular ion pairing (E21–K3–E48–R66, K7–D25, and K63–E52) than *Bs*-CspB, which contributes to the enhanced thermostability of *Ta*-Csp. Hyperthermophilic *Tm*-Csp with a T_m (~80 °C) has a peripheral ion cluster (D20–R2–E47–K63), two salt bridges (K6–D24 and H61–E49), and 12 hydrophobic core

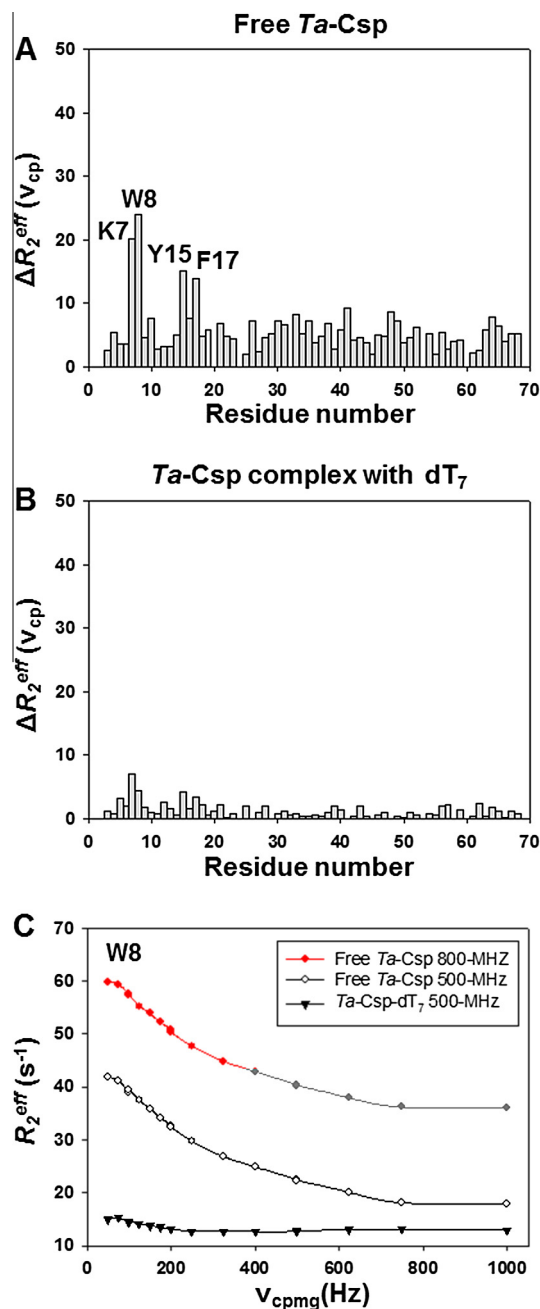


Fig. 4. Chemical exchanges in (A) free *Ta*-Csp and (B) *Ta*-Csp-dT₇. The value ΔR_2^{eff} (v_{cp}) was determined from the difference in measured relaxation rates, R_2^{eff} (v_{cp}, 50 Hz) – R_2^{eff} (v_{cp}, 1000 Hz). (C) Fitted data of W8 at the nucleic acid binding site.

residues [15], similar to *Ta*-Csp. Furthermore, the thermophilic *Ta*-Csp has five β-strands—β₁ (K2–N10), β₂ (Y15–Q20), β₃ (D25–H29), β₄ (H46–G54) and β₅ (P60–R67)—like all other Csps. Compared to hyperthermophilic *Tm*-Csp, *Ta*-Csp has shorter β₂ (G13–K19) and β₃ (D24–W29) strands, resulting in longer loop 2 and loop 3. These results imply that variations in β-strand length can play a critical role in determining the thermal stability of thermophilic Csps.

Internal motions have been previously reported in ¹⁵N relaxation studies of unliganded *Ec*-CspA, *Bs*-CspB, *Lm*-CspA and the *Bs*-CspB-dT₇, *Lm*-CspA-dT₇ complexes [9,16,30]. The binding of dT₇ to all Csps has been shown to reduce local dynamics and increase protein rigidity. The dT₇ binding introduces structural rigidity in the *Ta*-Csp structure, especially at the surface loop regions, as

evident by the increase in hNOE values upon dT₇ binding to *Ta*-Csp. In particular, the K13 (0.54 → 0.77) residue in loop 1 and K58 (0.60 → 0.77) residue in loop 4 showed dramatic increases in hNOE values in the dT₇-bound *Ta*-Csp compared with free *Ta*-Csp. These basic residues form contacts with the backbone phosphates of the DNA thereby contributing to the rigidity of *Ta*-Csp-dT₇ [16]. Compared to *Ta*-Csp, psychrophilic *Lm*-Csp is much more flexible and in particular loop3 is very flexible with extremely low hNOE values.

In *Ta*-Csp, high ΔR_2^{eff} values for residues exclusively in the β 1– β 2 surface region (e.g., K7, W8, Y15, and F17) imply that this nucleic acid binding site in free *Ta*-Csp are in a dynamic equilibrium between different conformers, while other regions of the molecule display high rigidity without microseconds to milliseconds motions. In thermophilic *Ta*-Csp, salt bridges in the β 4 and β 5 regions (E21–K3–E48–R66 and K63–E52) connect the two strands, conflating their conformational states. In contrast, in the case of psychrophilic *Lm*-CspA, slow motions appeared in the overall structure [16]. Given that there are only two salt bridges (E2–R20 and K7–D25) between the loop regions in *Lm*-CspA, there are millisecond to microsecond time scale motions in the broader region containing the F15, F17, H29, and R56 residues [16].

In this study, we revealed the thermostability and the dynamic properties of thermophilic Csp using NMR spectroscopy for the first time. We conclude that a number of salt bridges, the length of β -strands, and hydrophobic core packing are likely key factors in determining the thermostability of *Ta*-Csp. Furthermore, spin relaxation data revealed that nucleic acid binding stabilized the native conformations of *Ta*-Csp that are found in a dynamic equilibrium in the unbound state. In a free state, aromatic rings of W8, Y15, and F17 are exposed to the solvent and is in conformational exchange between the conformers while in bound state, *Ta*-Csp has dT₇ binding stabilized the major bound conformations. ¹⁵N relaxation dispersion experiments showed that slow exchange occurs only at the β 1– β 2 surface region with an average conformational exchange rate constant of $674 \pm 55.3 \text{ s}^{-1}$. As shown in *Lm*-Csp, psychrophilic Csp should be more flexible to accommodate nucleic acid at extremely low temperature than thermophilic Csp. In contrast, thermophilic *Ta*-Csp has a more rigid structure and may not need high structural flexibility to accommodate nucleic acids upon cold shock.

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